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## 4,4'-Bis(4-iodophenylaminomethyl)azoxybenzene: hydrogen-bonded sheets built from C—H···O and C—H··· $\pi$ (arene) hydrogen bonds

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Borohydride reduction of *N*-(4-nitrobenzylidene)-4-iodoaniline has yielded the title compound, 1,2-bis[4-(4-iodophenylaminomethyl)phenyl]diazene 1-oxide,  $C_{26}H_{22}I_2N_4O$ . The molecules lie across centres of inversion in  $P2_1/c$ , with the azoxy O atom disordered over two sites, each having an occupancy of 0.5. The molecules are linked into sheets by a combination of  $C-H\cdots O$  and  $C-H\cdots \pi$ (arene) hydrogen bonds.

#### Comment

Reduction of N-(nitrobenzylidene)iodoanilines, (a), using sodium borohydride, generally yields the corresponding N-(nitrobenzyl)iodoanilines (b), and six of the isomeric compounds (b) have so far been structurally characterized (Glidewell *et al.*, 2002, 2003). However, the analogous reduction of N-(4-nitrobenzylidene)-4-iodoaniline yielded not the expected 4-iodo-N-(4-nitrobenzyl)aniline but instead the title compound, (I), a further reduction of the nitrobenzylaniline presumed to have been the initial product of reduction.

Molecules of (I) lie across centres of inversion in space group  $P_{2_1/c}$ , with the reference molecule selected to lie across the inversion centre at  $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$ . The azoxy O atom is thus disordered over two sites, each with an occupancy of 0.5. The molecular conformation can be defined in terms of four independent dihedral angles (Fig. 1 and Table 1). The C11– C16 inner ring is nearly coplanar with the central azoxy unit, as is typically found in azoxybenzenes (Ejsmont, Domański *et*  *al.*, 2000; Ejsmont, Broda *et al.*, 2002), and the terminal part of the molecule, comprising the -C11-C17-N1-C21- unit and the iodinated outer ring, is also nearly planar. However, the C11-C16 ring and terminal fragment are far from being coplanar, as shown by the C12-C11-C17-N1 torsion angle.



The molecules are linked by  $C - H \cdots O$ and  $C-H\cdots\pi$ (arene) hydrogen bonds (Table 2). Because of the partial occupancy of the O-atom sites, we first discuss the effect of the C–H··· $\pi$ (arene) hydrogen bonds. Atom C16 at (x, y, z), which lies in the molecule centred at  $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$ , acts as hydrogen-bond donor to the C21–C26 ring at (x, y, 1+z), which lies in the molecule centred at  $(\frac{1}{2}, \frac{1}{2}, \frac{3}{2})$ , and propagation of this interaction by inversion and translation generates a molecular ladder running parallel to the [001] direction (Fig. 2). This ladder is reinforced by  $C-H \cdots O$  hydrogen bonds. Atom C12 at (x, y, z), which is part of the molecule centred at  $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$ , acts as hydrogen-bond donor to O14 at (x, x)y, -1 + z), which is part of the molecule centred at  $(\frac{1}{2}, \frac{1}{2}, -\frac{1}{2})$ . Since each molecule contains only one O atom, each molecule can accept only one  $C-H \cdots O$  hydrogen bond of this type, and a given pair of molecules, related by translation along the [001] direction, may in fact be linked by zero, one or two C- $H \cdots O$  hydrogen bonds, although on average there will be one such bond between each adjacent pair. The occurrence of exactly one C-H···O hydrogen bond between each pair of adjacent molecules would require perfect correlation of the Oatom site occupancies within a given ladder, although there need be no correlation between neighbouring ladders.

A similar, although more weakly bonded, ladder is generated along the [101] direction. Atom C25 at (x, y, z), in the molecule centred at  $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$ , acts as hydrogen-bond donor to the inner ring at (-1 + x, y, -1 + z), which forms part of the molecule centred at  $(-\frac{1}{2}, \frac{1}{2}, -\frac{1}{2})$ , while C17 at (x, y, z) acts as donor, *via* H17*B*, to O14, also at (-1 + x, y, -1 + z). Propagation of these interactions by inversion and translation generates the [101] ladder (Fig. 3), and the combination of ladders along [001] and [101] generates a sheet parallel to (010). Two such sheets run through each unit cell, in the domains 0.28 < y < 0.72 and -0.22 < y < 0.22, but there are no direction-specific interactions between adjacent sheets. In particular, there are no aromatic  $\pi$ - $\pi$ -stacking interactions, despite the number of aryl rings present. Note that the amino N atom plays no role in the supramolecular aggregation.

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#### Figure 1

The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are shown at the 30% probability level. Atom O14 has an occupancy of 0.5 and atoms marked with the suffix 'a' are at the symmetry position (1 - x, 1 - y, 1 - z).



#### Figure 2

A stereoview of part of the crystal structure of (I), showing the formation of a molecular ladder along [001].



#### Figure 3

A stereoview of part of the crystal structure of (I), showing the formation of a molecular ladder along [101].

#### **Experimental**

A sample of (I) was obtained by reduction of the corresponding N-(benzylidene)aniline using a fivefold molar excess of NaBH<sub>4</sub> in refluxing methanol for 1 h. After work-up, crystals of (I) suitable for single-crystal X-ray diffraction were grown by the slow evaporation of a solution in ethanol.

#### Crystal data

$C_{26}H_{22}I_2N_4O$	$D_{\rm r} = 1.803 {\rm Mg} {\rm m}^{-3}$
$M_r = 660.28$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 4439
a = 5.6959(7)  Å	reflections
b = 30.061 (3)  Å	$\theta = 2.7 - 33.0^{\circ}$
c = 7.2017 (8)  Å	$\mu = 2.61 \text{ mm}^{-1}$
$\beta = 99.526 \ (2)^{\circ}$	T = 291 (2)  K
$V = 1216.1 (2) \text{ Å}^3$	Plate, colourless
Z = 2	$0.40\times0.30\times0.02~\mathrm{mm}$

#### Data collection

Bruker SMART 1000 CCD area-	4439 independent reflections
detector diffractometer	3092 reflections with $I > 2\sigma(I)$
$\varphi$ and $\omega$ scans	$R_{\rm int} = 0.038$
Absorption correction: multi-scan	$\theta_{\rm max} = 33.0^{\circ}$
(SADABS; Bruker, 2000)	$h = -8 \rightarrow 8$
$T_{\min} = 0.403, T_{\max} = 0.946$	$k = -39 \rightarrow 46$
12 146 measured reflections	$l = -11 \rightarrow 10$

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0409P)^2$
$R[F^2 > 2\sigma(F^2)] = 0.065$	+ 3.3092 <i>P</i> ]
$wR(F^2) = 0.156$	where $P = (F_o^2 + 2F_c^2)/3$
<i>S</i> = 1.13	$(\Delta/\sigma)_{\rm max} = 0.001$
4439 reflections	$\Delta \rho_{\rm max} = 1.07 \text{ e } \text{\AA}^{-3}$
154 parameters	$\Delta \rho_{\rm min} = -1.19 \mathrm{e}\mathrm{\AA}^{-3}$
H-atom parameters constrained	

#### Table 1 S al

Selected	torsion	angles	C	)

C13-C14-N14-N14 <sup>i</sup>	17.4 (9)	C11-C17-N1-C21	-170.8 (5)
C12-C11-C17-N1	51.7 (8)	C17-N1-C21-C22	-173.5 (6)

Symmetry code: (i) 1 - x, 1 - y, 1 - z.

### Table 2

Hydrogen-bonding geometry (Å, °).

Cg1 and Cg2 are the centroids of rings C11-C16 and C21-C26, respectively.

$D - H \cdot \cdot \cdot A$	D-H	$H \cdots A$	$D \cdots A$	$D - H \cdots A$
C12-H12O14 <sup>ii</sup>	0.93	2 24	3 078 (9)	150
$C17 - H17B \cdots O14^{iii}$	0.95	2.45	3.380 (10)	162
$C16-H16\cdots Cg2^{iv}$	0.93	2.78	3.536 (5)	139
$C25-H25\cdots Cg1^{iii}$	0.93	2.97	3.631 (5)	130

Symmetry codes: (ii) x, y, z - 1; (iii) x - 1, y, z - 1; (iv) x, y, 1 + z.

Compound (I) is monoclinic and the space group  $P2_1/c$  was uniquely assigned from the systematic absences. H atoms were treated as riding atoms, with C-H distances of 0.93 (aromatic) and 0.97 Å (CH<sub>2</sub>), and N-H distances of 0.86 Å. Refinement of the siteoccupancy factor for atom O14 gave a value of 0.52 (3); this factor was thereafter fixed at 0.50.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS*86 (Sheldrick, 1990); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL*97 and *PRPKAPPA* (Ferguson, 1999).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: SK1624). Services for accessing these data are described at the back of the journal.